



**DAH-003-1014004**

Seat No. \_\_\_\_\_

**B. Sc. (Sem. IV) Examination**

**April - 2022**

**Chemistry : C-401**

*(Old Course)*

**Faculty Code : 003**

**Subject Code : 1014004**

Time :  $2\frac{1}{2}$  Hours]

[Total Marks : 70

- Instructions :**
- (1) This question paper contains five questions.
  - (2) All questions carry 14 marks each and figures to the right indicate full marks.
  - (3) Write sub questions (a), (b), (c) and (d) of particular question together.

- 1 (a) Answer the following questions : 4
- (1) Write the formula of Zeise's salt.
  - (2) Which metal is associated in the structure of chlorophyll ?
  - (3) Which heterocyclic ring is present in Porphyrins ?
  - (4) Define organometallic compound.
- (b) Answer any **one** : 2
- (1) Give two preparation of organolithium compounds.
  - (2) Explain myoglobin.
- (c) Answer any **one** : 3
- (1) Explain classification of organometallic compounds based on M-C bond.
  - (2) Explain importance of chlorophyll.
- (d) Answer any **one** : 5
- (1) Explain the structure of trimethyl aluminium (dimer).
  - (2) Explain the structure and role of Hemoglobin in biological system.

- 2 (a) Answer the following questions : 4
- (1) Complete the reaction  $XeF_2 + H_2 \rightarrow$
  - (2) Name **one** noble gas elements make a number of compounds ?
  - (3) Give the structure of Methylene group.
  - (4) Give the reaction of Ethylacetoacetate with HCN.
- (b) Answer any **one** : 2
- (1) Give preparation of  $KrF_2$ .
  - (2) Explain Keto-Enol tautomerism with example.
- (c) Answer any **one** : 3
- (1) Explain Hybridization and structure of  $XeF_4$ .
  - (2) Discuss principle of claisen condensation.
- (d) Answer any **one** : 5
- (1) Explain uses of noble gases.
  - (2) Give synthesis of valeric acid and levulenic acid from EAA.
- 3 (a) Answer the following questions : 4
- (1) What is carbonyl group ?
  - (2) Give IUPAC name of Butyraldehyde.
  - (3) Give structure of Acetic anhydride.
  - (4) Write structure of Acetamide.
- (b) Answer any **one** : 2
- (1) Give synthesis of Aldehyde from acid chloride.
  - (2) Explain acidity of carboxylic acid.
- (c) Answer any **one** : 3
- (1) Explain Wolff Kishner and Clemmensen reduction of Ketone.
  - (2) Explain HVZ reaction.

- (d) Answer any **one** : 5
- (1) Explain nucleophilic addition elimination reaction of ammonia derivatives with aldehydes.
  - (2) What is Trans esterification ? Explain in detail with mechanism.
- 4 (a) Answer the following questions : 4
- (1) Write the structural formula of witting reagent.
  - (2) In aldol condensation which product is obtain ?
  - (3) What is the CGS unit for dipole moment ?
  - (4) Define surface tension.
- (b) Answer any **one** : 2
- (1) Explain principle of witting reaction.
  - (2) Write short note on Parachor.
- (c) Answer any **one** : 3
- (1) Give mechanism of Perkin Reaction.
  - (2) What is Refractive index ? Explain Molar refractivity of solutions and solids.
- (d) Answer any **one** : 5
- (1) Explain Beckmann rearrangement with mechanism.
  - (2) Explain Ostwald's viscometer.
- 5 (a) Answer the following questions : 4
- (1) Name the process in which pressure remains constant.
  - (2) Define specific heat with equation.
  - (3) Define Enthalpy.
  - (4) What is Close system ?
- (b) Answer any **one** : 2
- (1) Write note on type of processes.
  - (2) Write two statement of first law of thermodynamics.

- (c) Answer any **one** : **3**
- (1) Write note on advantages and limitation of thermodynamics.
  - (2) Derive relationship between pressure and volume for adiabatic process of ideal gas.
- (d) Answer any **one** : **5**
- (1) Derive  $C_p - C_v = R$ .
  - (2) State Kirchoff's law and derived it.
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